

Errata

The thermodynamics of ionisation of glycine in methanol + water mixtures and the determination of single ion thermodynamics, by S.K. Chakravarty and S.C. Lahiri, *Thermochim. Acta*, 99 (1986) 243.

Page 244, fifth line from the bottom should read:
amino acid (5 ml of 3.03–6.33 M/10) was taken in an appropriate solvent,
to

Stoichiometry, thermodynamic functions and stability of some charge transfer complexes, by H.S. Randhawa, S. Rani, R. Sachdeva and S.K. Suri, *Thermochim. Acta*, 105 (1986) 303.

Table 2, page 307, the Δ value for naphthalene ($\phi\text{-CH=N-}\phi\text{-NO}_2$) should read

1 : 2 : 1 : 1 : 3

that for phenanthrene ($\phi\text{-CH=N-}\phi\text{-NO}_2$) should read

1 : 1 : 11 : 2

Table 3, page 310, the relative stability of the molecular complex between $\text{HO-}\phi\text{-CH=N-}\phi$ and naphthalene should read

2 : 3 > 3 : 2 > 3 : 1 = 7 : 3 > 1 : 4

that for $\phi\text{-CH=N-}\phi\text{-NO}_2$ and anthracene should read

7 : 3 > 1 : 3

that for $\phi\text{-CH=N-}\phi\text{-NO}_2$ and phenanthrene should read

7 : 3 > 4 : 1 > 1 : 3 = 3 : 17

that for $\phi\text{-CH=N-}\phi\text{-Cl}$ and phenanthrene should read

1 : 4 > 9 : 1 = 3 : 2 = 3 : 1